

0040-4039(93)E0395-Z

Enantioselective Synthesis of 5-LO Inhibitors Using a Gulofuranose Auxiliary¹

John C. Rohloff,* Thomas V. Alfredson[†] and Martin A. Schwartz[‡]

*Institute of Organic Chemistry and [†]Institute of Analytical Research, Syntex Discovery Research, Palo Alto, CA 94303

> [‡]Department of Chemistry, Florida State University, Tallahassee. FL 32306

Abstract: The pure R-(+)-enantiomer of 5-lipoxygenase inhibitor Zileuton was prepared by diastereoselective methyl Grignard addition to an aldonitrone bearing a D-gulofuranose-derived chiral auxiliary. Addition of the Lewis acid trimethylaluminum leads to a reversal of the alkylation stereochemistry and the potent pyrido analogue, R-(+)-RS-27871 was prepared in this way from an L-gulofuranose-derived nitrone.

Zileuton 1 is a racemic N-(1-arylethyl)-N-hydroxyurea inhibitor of mammalian 5-lipoxygenase (5-LO) which is currently in clinical development by Abbott Laboratories as a treatment for asthma.² The Syntex compound, RS-27871, 2 is an optically active pyrido analogue, bearing the $R-(+)$ -configuration, which shows higher potency and significantly longer metabolic half-life in vitro.³ Several routes to racemic 1 have been described in the literature but only one enantioselective synthesis has appeared, a lengthy chiral pool approach starting from L-lactic acid.^{2c} The lack of general enantioselective methodology for this class of compounds has been noted.^{2e}

Schwartz and Hu have recently disclosed a novel synthesis of chiral benzylic amines and hydroxylamines based upon Grignard alkylation of nitrones bearing a 2,3:5,6-di-O-isopropylidene-gulofuranose auxiliary.⁴ A particular advantage of the sugar gulose is that both of its enantiomers are readily prepared, and commercially available, in the form of the gulono-1,4-lactones. Related precedents were found in the preparation and diastereoselective cycloaddition reactions of sugar-derived aldonitrones which have been extensively studied by Vasella.⁵ In addition, the utility of chelation-controlled stereoselective Grignard additions to amino alcohol-derived nitrones has been amply demonstrated by Coates.⁶

The synthesis of enatiomerically pure $R-(+)$ -Zileuton was readily accomplished using the gulose auxiliary (Scheme 1). D-Gulono-1,4-lactone⁷ 3 was protected as its diisopropylidene derivative (acetone, 2,2dimethoxypropane, TsOH),⁸ reduced to the lactol (Dibal, tol. -78 $^{\circ}$ C)⁹ and converted to a glassy 6:5 mixture of aldoximes 4 (hydroxylamine+HCl, sodium bicarbonate, aq. MeOH) in 42% overall yield. Thianaphthene 5a was formylated (1.4 eq. n-BuLi, THF, -78 $^{\circ}$ C; 4 eq. DMF)¹⁰ to give aldehyde 6a in 82% yield and this was

condensed with equimolar 4 in refluxing toluene under Dean-Stark conditions to afford the highly crystalline nitrone 7 a^{11} in 78% yield. The β -configuration of the anomeric center was assigned based on observation of J_1 - 2 ⁻ \sim O Hz in the ¹H-NMR^{5b} and the Z-nitrone stereochemistry was supported by observation of a strong NOE between the benzylic and Cl' hydrogens.

Treatment of a slurry of the nitrone 7a with methylmagnesium bromide at 0° C (1.5 eq. ethereal MeMgBr, CH₂Cl₂, 1 h) gave a clear solution from which the sole monoalkylation product, $8a^{12}$ (>99% de), was isolated in 61% yield following silica gel chromatography. Acid cleavage of **8a** (1N HCl, MeOH, 0°C, 12h) afforded the hydroxylamine 9a which was carbamoylated (TMSNCO, THF, 23°C) without purification to give $R-(+)$ -1 in 72% yield, after crystallization from CH₂Cl₂/MeOH/hexane.¹³

Preparation of optically pure RS-27871 2 was initially found to be more difficult. Thieno[2,3-b]pyridine 5b¹⁴ was formylated¹⁵ and condensed with 4, as above, to give nitrone 7b¹⁶ in 80% yield. Alkylation at 0° C with MeMgBr in CH₂Cl₂ gave a 64% yield of a mixture of monoalkylation products, with a disappointing 30% de favoring the desired R-epimer **8b. The** lower diastereoselectivity in this case presumably resulted from perturbation of the structured magnesium chelate by the pyrido ligand. At -78°C no reaction occurred (20 MeMgBr, 40h, CH₂Cl₂ or THF) due to the insolubility of **7b** (or **7a**) at low temperature.

Alternative organometallic reagents were screened in an attempt to raise the diastereoselectivity and this led to a serendipitous discovery. Addition of trimethylaluminum (1.2 eq 1M in hexane) to a CH₂Cl₂ slurry of **7b** gave no alkylation but instead formed a stable, soluble complex. The complex **remained** fully dissolved even on cooling to -78°C and Grignard addition at that temperature (1.5 eq ethereal MeMgBr, 0.5 h) unexpectedly afforded, as the major product, the S-epimer 10^{17} (63% yield, 82% de) revealing a *reversal of the alkylation* facial selectivity from the uncomplexed experiments.¹⁸ Repetition of the sequence starting from L-gulono-1,4lactone **ent-3,19** using aluminum-mediated Grignard addition, afforded the desired R-epimer, **ent-IO,** in similar overall yield and de. Acid cleavage of **ent-10** gave **9b,** which upon carbamoylation afforded pure 220 in 40% yield, after removal of a small amount of racemic material by crystallization.

Comparison with two closely related sugar auxiliaries **implicated** gulose's CS-oxy substituent as a key stereocontrol element, **despite** its distance from the reaction center. Commercially available 2,3:5,6-di-0 isopropylidene-a-D-mannofuranose was readily converted into its oximes and condensed with aldehyde Sb (toluene, 110° C, 40 h) to give nitrone 11^{21} in 65% yield. In similar fashion, L-erythronic γ -lactone was converted into nitrone 12.22 **Despite the** obvious structural homologies of **11** and **12 with 7b (or more** specifically with **ent-7b),** methyl Grignard addition to these former nitrones, either with or without trimethylaluminum, proceeded with poor diasteteoselectivity. Thus, Schwartz's identification of the **gulofuranose** auxiliary appears to be a key observation, with the trimethylaluminum-mediated reaction being a useful complement to the basic method in some cases. 23

Ackhowledgment: The authors thank Mr. Doug **Wren for the preparation of 5b and** Mr. David Repke for supplying a reference sample of racemic Zileuton 1.

References:

- Contribution **X888 from the Institute of Orgauic Chemisuy, Syntex Discovery Research, Palo Alto. CA. This paper is** 1. dedicated to Dr. John A. Edwards on the occasion of his retirement as Institute Director.
- $2.$ (a) Batt, D. G. Progress In Medicinal Chemistry 1992; Ellis, G. P.; Luscombe, D. K., Eds.; Elsevier, Amsterdam, Vol.
29. pp. 2-46. (b) Basha, A.; Brooks, D. W. J. Org. Chem. 1993, 58, 1293-4. and references cited therein. N.; Kolasa, T. *Tetrahedron Lett.* 1992, 33, 2629-2632. (d) Abraham, W. M.; Ahmed, A.; Cortes, A.; Sielczak, M. W.;
Hinz, W.; Bouska, J.; Lanni, C.; Bell, R. L. *Eur. J. Pharmacol.* 1992, 217, 119-26. (e) Garigipati, R. S. **E.; Erhard, K. F.; Adams, 1. L.** *Tetrahedron Len.* **1993,34.553740.**
- Unpublished results from Syntex Discovery Research, Palo Alto, CA 94303. 3.
- (a) Hu, X.; Schwartz, M. A. Book of Abstracts , 204h ACS National Meeting, Washington, D. C., 4.
- **5. 1992.** Abstract ORGN 5. (b) Schwartz, M. A.; Hu, E. X.; Savidakis, M. C. J. Org. Chem., submitted for publication. a) Vasella, A.; Voeffray, R.; Pless, J.; Huguenin, R.; *Helv. Chim. Acta* 1983, 66, 1241-6. (b)
- **Vasella, A.** *ibid 1977.60. 1273-95.*
- б. Chang, Z.-Y.; Coates, R. M. J. Org. Chem. 1990, 55, 3464-74, 1275-95.
- *S: Karabiios, I.* V. Org. *Synth.CoU. Vol.iV 1963, 506-8.*
- *8. Fleet, G.* **W. J.: Ramsden, N. G.; Witty, D. R. Terruhedron 1989.45.319-26.**
- 9. (a) Rosen, T.; Taschner, M. J.; Heathcock, C. H. J. Org. Chem. 1984, 49, 3994-4003. (b) Iida, K.; Kasahara, K.; Kibayashi, C. J. Am. Chem. Soc. 1986, 108, 4647-4648.
- Shirely, D. H.; Danzig, M. J. J. Am. Chem. Soc. 1952, 74, 2935-6. 10.
- (7a): mp 244-6°C (EtOAc/tol); [α]_D -14° (c 1.1, CH₂Cl₂); IR (KBr), 1563 cm⁻¹; FAB MS, m/z 420 (MH⁺), 243, 285, 11. 127; ¹H NMR (CDCl3, 300 MHz) δ 8.14 (s, 1 H, N=CH), 7.92-7.80 (m, 2 H, ArH), 7.76 (s, 1H, Ar3H), 7.46-7.38 (m, 2 H, ArH), 5.65 (s, 1 H, C₁·H), 5.37 (d, J = 6.1 Hz, 1 H, C₂·H), 4.94 (dd, J = 4.3, 6.1 Hz, 1 H, C₃·H), 4.65 (dd, J = 4.3, 8.1 Hz, 1 H, C4·H), 4.39 (ddd, J = 6.6, 7.3, 8.1, 1 H, C5·H), 4.25 (dd, J = 6.6, 8.5 Hz, C6·H), 3.73 (dd, J = 7.3, 8.5 Hz, C_6 H'), 1.52 (s, 3 H, Me), 1.46 (s, 3 H, Me), 1.41 (s, 3 H, Me), 1.34 (s, 3 H, Me). Anal. Calcd for C_2 1H₂₅NO₆S: C, 60.13; H, 6.01; N, 3.34. Found: C, 60.04; H, 5.76; N, 3.51.
- (8a): mp 87°C (foam); [α]_D +38° (c 0.6, CH₂Cl₂); IR (KBr), 3440 cm⁻¹; MS m/z 435 (M⁺'), 377, 319, 161; ¹H NMR 12. (CDCl3, 300 MHz) 8 7.80-7.68 (m, 2 H, ArH), 7.35-7.25 (m, 2 H, ArH), 7.20 (s, 1 H, C3H), 5.12 (s, 1 H, OH), 4.98 (d, $J = 6.1$ Hz, 1 H, C₂·H), 4.74 (dd, $J = 4.2$, 6.1 Hz, 1 H, C₃·H), 4.73 (s, 1 H, C₁·H), 4.70 (q, $J = 6.6$ Hz, 1 H, ArCHN), 4.36-4.18 (m, 3 H, C_{4} '5'6'H), 3.68 (dd, J = 6.5, 8.2 Hz, 1 H, C_{6} 'H'), 1.50 (d, J = 6.6 Hz, 3 H, NCMe), 1.49 (s, 3 H, Me), 1.48 (s, 3 H, Me), 1.41 (s, 3 H, Me), 1.32 (s, 3 H, Me). Anal. Calcd for C22H29NO6S: C, 60.67; H, 6.71; N, 3.22. Found: C, 60.72; H, 6.56; N, 3.33.
- ((R)-(+)-1): mp 155-7°C (CH₂Cl₂/MeOH/hexane); [α]_D +46.8° (c 0.349, MeOH), [lit.^{2c} [α]_D +50.3° (c 0.35, MeOH)]; 13. IR (KBr), 3459, 3276, 1676 cm⁻¹; ¹H NMR (5% DMSO-d₆/CDCI3, 300 MHz) δ 8.83 (s, 1 H, OH), 7.78-7.69 (m, 2 H, ArH), 7.33-7.24 (m, 3 H, ArH), 5.77 (q, J = 6.9 Hz, 1 H, ArCH), 5.50 (bs, 2 H, NH₂), 1.65 (d, J = 6.9 Hz, 3 H, Me). Anal. Calcd for C11H12N2O2S: C, 55.91; H, 5.12; N, 11.86. Found: C, 56.05; H, 5.10; N, 11.73. Chiral purity of >99.6% ee was verified by HPLC upon comparison with a racemic reference sample^{2b} (Chiralcel OD-R, 4.6 mm x 25 cm, 35°C, 75% 0.5 M NaClO4/25% CH₃CN, $\lambda = 234$ nm, flow 1 mL/min.).
- Klemm, L. H.; Klopfenstein, C. E.; Zell, R.; McCoy, D. R.; Klemm, R. A. J. Org. Chem. 1969, 34, 347-54. 14.
- Merrill, R. E.; Klemm, L. H. J. Heterocyclic Chem. 1974, 11, 355-361. 15.
- (7b): mp 254-5°C (EtOAc/tol); [α]p -12.5° (c 1, CH₂Cl₂); IR (KBr), 1552 cm⁻¹; ¹H NMR (DMSO-d6, 300 MHz) δ 16. 8.79 (s, 1 H, N=CH), 8.64 (dd, J = 1.6, 4.6 Hz, 1 H, C₆H), 8.30 (dd, J = 1.6, 8.2 Hz, 1 H, C₄H), 7.95 (s, 1 H, C₃H), 7.46 (dd, J = 4.6, 8.2 Hz, 1 H, C5H), 5.94 (s, 1 H, C1H), 5.17 (d, J = 6.0 Hz, 1 H, C2H), 5.02 (dd, J = 4.0, 6.0 Hz, 1 H, C₃^H), 4.61 (dd, J = 4.2, 8.2 Hz, 1 H, C₄^H), 4.19-3.79 (m, 3 H, C₅^t, 6^H), 1.46 (s, 3 H, Me), 1.36 (s, 3 H, Me), 1.30 (s, 6 H, 2 x Me). Anal. Calcd for C20H24N2O6S: C, 57.13; H, 5.75; N, 6.66. Found: C, 57.27; H, 5.68; N, 6.66.
- (10:8b, 91:9 crystalline mixture): mp 161-2°C (EtOAc/hex); $[\alpha]_D+110^{\circ}$ (c 1, CH₂Cl₂); IR (KBr), 3430 cm⁻¹; MS m/z 17. 436 (M⁺'), 378, 162; Anal. Calcd for C₂₁H₂₈N₂O₆S: C, 57.78; H, 6.46; N, 6.42. Found: C, 57.92; H, 6.47; N, 6.55. (10): ¹H NMR (CDCl3, 300 MHz) δ 8.47 (dd, J = 1.4, 4.7 Hz, 1 H, C₆H), 7.94 (dd, J = 1.4, 8.0 Hz, 1 H, C4H), 7.24 (dd, $J = 4.7, 8.0$ Hz, 1 H, C5H), 7.21 (s, 1 H, C3H), 6.26 (bs, 1 H, OH), 4.97 (d, $J = 6.1$ Hz, 1 H, C2 H), 4.84 (s, 1 H, C_1 ^H), 4.78 (dd, J = 4.5, 6.1 Hz, 1 H, C₃^H), 4.71 (g, J = 6.5 Hz, 1 H, ArCHN), 4.38 (dd, J = 4.5, 8.1 Hz, C4[·]H), 4.30-4.17 (m, 2 H, C5'6'H), 3.70 (dd, J = 6.6, 8.0 Hz, C6'H), 1.64 (d, J = 6.5 Hz, 3 H, NCMe), 1.48 (s, 3 H, Me), 1.41 (s, 3 H, Me), 1.38 (s, 3 H, Me), 1.25 (s, 3 H, Me).
- Methyl Grignard addition to 7a in the presence of trimethylaluminum gave 8a in 20% de. 18.
- Hubschwerlen, C. Synthesis 1986, 962-4, ref. 6. 19.
- (2): mp 158-60°C (McOH/EtOAc); [alp +41° (c 0.2, MeOH); IR (KBr), 3475, 3150, 1693 cm⁻¹; ¹H NMR (DMSO-d₆, 20. 300 MHz) δ 9.32 (s, 1 H, OH), 8.49 (dd, J = 1.6, 4.7 Hz, 1 H, C₆H), 8.16 (dd, J = 1.6, 8.0 Hz, 1 H, C₄H), 7.38 (dd, J = 4.7, 8.0 Hz, 1 H, C5H), 7.26 (d, J = 0.9 Hz, 1 H, C3H), 6.52 (bs, 2 H, NH2), 5.58 (dq, J = 0.9, 7.0 Hz, 1 H, ArCHN), 1.52 (d, J = 7.0 Hz, 3 H, Me). Anal. Calcd for C10H11N3O2S: C, 50.61; H, 4.67; N, 17.70. Found: C, 50.64; H, 4.68; N, 17.74. Chiral purity of 96% ee was determined by HPLC (Chiral AGP, 4 mm x 10 cm, 23°C, 92% 10 mM KH2PO4/ 8 % MeOH, $\lambda = 234$ nm, flow 1 mL/min.).
- (11): mp 248-51°C (EtOAc/tol); [α]D + 58° (c 1, CH2Cl2); IR (KBr), 1553 cm⁻¹. Anal. Calcd for C20H24N2O6S: C, 21. 57.13; H, 5.75; N, 6.66. Found: C, 56.69; H, 5.63; N, 6.61.
- (12): mp 210-3°C (EtOAc/tol); [α]p + 145° (c 0.6, CH₂Cl₂); IR (KBr), 1552 cm⁻¹. Anal. Calcd for C15H16N₂O4S: C, $22.$ 56.24; H, 5.03; N, 8.74. Found: C, 56.34; H, 5.18; N, 8.96.
- 23. Other examples of Lewis Acid-mediated reversal of nitrone alkylation facial selectivity have recently been reported; Dondini, A.; Franco, S.; Merchan, F. L.; Merino, P.; Tejero, T. Tetrahedron Lett. 1993, 34, 5475-5478, 5479-82.

(Received in USA 3 November 1993; accepted 10 December 1993)

1014